

Name	Isotope Notation	Atomic Number	Mass Number	Number of Protons	Number of Neutrons	Number of Electrons
	$^{14}\text{N}$					
	$^{35}\text{Cl}$					
	$^{39}\text{K}$					

### Answers

Name	Isotope Notation	Atomic Number	Mass Number	Number of Protons	Number of Neutrons	Number of Electrons
Nitrogen	$^{14}\text{N}$	7	14	7	7	7
chlorine	$^{35}\text{Cl}$	17	35	17	18	17
potassium	$^{39}\text{K}$	19	39	19	20	19